

Journal of Molecular Catalysis A: Chemical

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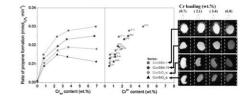
Articles

Piotr Michorczyk, Jan Ogonowski, Kamila Zeńczak

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Activity of chromium oxide deposited on different silica supports in the dehydrogenation of propane with CO_2 – A comparative study

▶ Chromium oxide dispersed on commercial silicas and siliceous mesoporous materials as a catalysts for dehydrogenation of propane with CO^2 (DHP-CO₂). ▶ Content of Cr redox species increases with the rise of specific surface area of the support. ▶ Propene formation rate is proportional to the number of Cr redox species in the catalysts. ▶ Presented in the fresh catalyst Cr(VI) species are reduced rapidly to Cr(II)/Cr(III) species at the beginning of the DHP-CO₂ process.



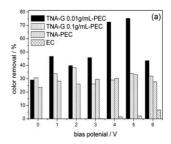
CATAL

Yajun Wang, Jie Lin, Ruilong Zong, Jun He, Yongfa Zhu

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Enhanced photoelectric catalytic degradation of methylene blue via TiO₂ nanotube arrays hybridized with graphite-like carbon

► Graphite-like carbon/TiO₂ nanotube array was synthesized by sucrose graphitization. ► Enhanced photoelectric catalytic activity under ultraviolet irradiation. ► Obvious photoelectric catalytic activity under visible irradiation ($\lambda > 450$ nm). ► Different degradation mechanisms of MB under ultraviolet and visible irradiation.

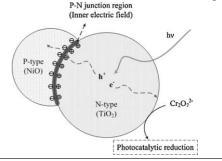


Young Ku, Chia-Nan Lin, Wei-Ming Hou

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Characterization of coupled NiO/TiO₂ photocatalyst for the photocatalytic reduction of Cr(VI) in aqueous solution

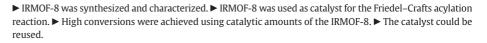
► The p-n junction mechanism is generated by coupling NiO and TiO₂ in sol-gel process. ► NiO/TiO₂ is fabricated for the photocatalytic reduction of Cr(VI). ► The p-n junction region can efficiently transfer the photogenerated holes. ► The presence of NiO retards the phase transformation and the growth of TiO₂.

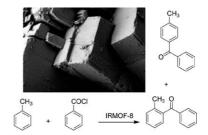


Lien T.L. Nguyen, Chi V. Nguyen, Giao H. Dang, Ky K.A. Le, Nam T.S. Phan

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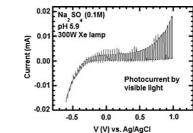
Towards applications of metal-organic frameworks in catalysis: Friedel-Crafts acylation reaction over IRMOF-8 as an efficient heterogeneous catalyst





D.P. Das, Niranjan Biswal, Satyabadi Martha, K.M. Parida

► CdS pillaring has decreased the band gap energy of ZTP to 2.81 eV. ► CdS pillaring has increased the surface area. ► CdS pillaring has shifted the adsorption edge to visible light. ► 15CdS-ZTP showed the best photoactivity towards the degradation of contaminants.



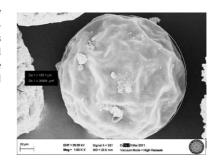
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Solar-light induced photodegradation of organic pollutants over CdS-pillared zirconium-titanium phosphate (ZTP)

Ali Kara, Beyhan Erdem

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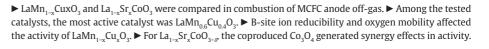
Synthesis, characterization and catalytic properties of sulfonic acid functionalized magneticpoly(divinylbenzene-4-vinylpyridine) for esterification of propionic acid with methanol ► We investigate the catalytic property of new material on the esterification of methyl propionate. ► m-Poly(DVB-4VP-SO₃H) is shown as active catalyst for propionic acid-methanol esterification. ► This new catalyst may be proposed as an alternative instead of commercial resins.

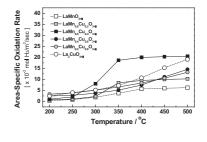


Hee-Jun Eom, Jun Ho Jang, Dae-Won Lee, Seongmin Kim, Kwan-Young Lee

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Catalytic combustion of hydrogen over $La_{1-x}Sr_xCoO_{3-\delta} + Co_3O_4$ and $LaMn_{1-x}Cu_xO_{3+\delta}$ under simulated MCFC anode off-gas conditions



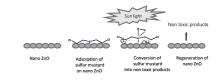


G.K. Prasad, P.V.R.K. Ramacharyulu, Beer Singh, K. Batra, Anchal R. Srivastava, K. Ganesan, R. Vijayaraghavan

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Sun light assisted photocatalytic decontamination of sulfur mustard using ZnO nanoparticles

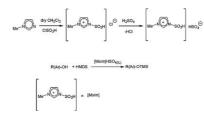
▶ Synthesis of ZnO nanoparticles of different sizes using sol gel method. ▶ Characterization of ZnO nanoparticles was done by XRD, IR, SEM and N₂ BET analysis. ▶ Advantage of ZnO is that, it absorbs over a larger fraction of solar spectrum. ▶ ZnO nanoparticles exhibited better efficiency towards sulfur mustard in Sun light. ▶ Photocatalysis aided faster decontamination of sulfur mustard on nano ZnO.



Nader Ghaffari Khaligh

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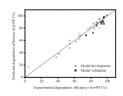
Preparation, characterization and use of 3-methyl-1sulfonic acid imidazolium hydrogen sulfate as an eco-benign, efficient and reusable ionic liquid catalyst for the chemoselective trimethylsilyl protection of hydroxyl groups ▶ Introducing the new catalyst (as ionic liquid) for organic transformations. ▶ Superiority of the catalyst with respect to the reported catalysts. ▶ Introducing an efficient new method for protection of hydroxyl groups. Generality of the method, high yields and very short reaction times. ▶ Application of solvent free conditions in the reaction.



Sheng-Peng Sun, Ann T. Lemley

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p-Nitrophenol degradation by a heterogeneous Fenton-like reaction on nano-magnetite: Process optimization, kinetics, and degradation pathways ▶ Degradation of p-NP in water by heterogeneous Fenton on nano-Fe₃O₄ at neutral pH. ▶ Parameter optimization by central composite design and response surface methodology. ▶ Primary reactive species, intermediate products and kinetics of p-NP degradation. ▶ An environmentally benign technology for remediation of p-NP and other contaminants.

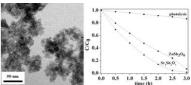


Wenjun Liu, Pingyong Lin, Hua Jin, Hun Xue, Yongfan Zhang, Zhaohui Li

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► Facile hydrothermal method to nanocrystalline ZnSb2O6 with large surface area. ► Nanocrystalline ZnSb₂O₆ shows photocatalytic activity for the degradation of dyes. ► ZnSb₂O₆ and Sr₂Sb₂O₇ show different activity due to different electronic structures. ► Second metal in ternary antinonates influence their electronic structure and activity.

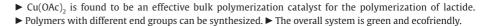
Nanocrystalline ZnSb₂O₆: Hydrothermal synthesis, electronic structure and photocatalytic activity



Ravikumar R. Gowda, Debashis Chakraborty

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Copper acetate catalyzed bulk ring opening polymerization of lactides



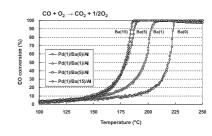


K. Tanikawa, C. Egawa

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Effect of barium addition over palladium catalyst for CO–NO–O, reaction

▶ Ba addition on Pd supported on Al_2O_3 improved both of CO oxidation and NO reduction. ▶ Ba addition caused the decrease for the activity of CO oxidation on Pd/CZ. ▶ Deactivation of CO oxidation is due to the reduction of oxygen supply from CZ to Pd. ▶ NO reduction is not affected by Ba on CZ under a stoichiometric CO-NO-O_2 condition.



Stefano Livraghi, Maria Cristina Paganini, Elio Giamello

► SO₂ interaction on oxide surface is affected by hydroxyl groups. ► EPR is the most suitable spectroscopic technique to evaluate radical species. ► Adsorption of SO₂ on oxide surfaces generates radicals. ► Different basicity of MgO, CaO and SrO plays a crucial role in reactivity.

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EPR study of electron trapping on partially hydroxylated alkali-earth oxides occurring during SO₂ disproportionation

